



11/10/15  
 Engineering  
 Physics  
 Color Center in Alkali Halides Symposium  
 Oct. 11-13, 1965

The four reels of tape are recorded at 3 3/4 speed on a dual track stereo recorder so that the four voice tracks are on each reel of tape.

The following index is keyed to the review papers and the slides presented by each speaker. The complete abstract of the review papers appears in the printed program.

### Reel 1 - Track 1

0-45	Empty	
46-	Theory of Optical Transitions in the Electron-Excess Centers by W. Beall Fowler (University of Illinois)	
	Discussion of lattice states	
120-	Computation of electronic properties	
145-	Criteria	
	1- position of the nuclei	
	2- proper consideration of electronic polarization	
	3- Pauli principle and ionic potentials	
170-	F centers and U centers are neutral centers	
185-	F centers are easiest to calculate.	
	Wave functions in absorption	
223-	Computation of energies and wave function of the F center	
	1- Common point - ion potential	Slide 1
260-	Extended ion potential	
274-	2- Semi-continuum potential of Simpson	Slide 2
300-	Constant energy	
330-	Comparison of models 1 and 2	
	1 is better for F band energy calculation	Slides 3, 4 & 5
370-	Higher excited states of the F center	
397-	Calculation of K band of F center	
425-	F center in emission	
	Lifetime of excited F center	Slide 6
		Slide 7
		Slide 8
495-	Relaxed states on conduction bands	Slide 9
528	3 distinct states of the F center band	
	1- tightly bound state - no polarization	
	2- K-band region - semi-continuum states characterized by a high frequency dielectric constant.	
	3- static dielectric constant with more diffuse excited states.	

540-	M centers	Slide 10
592-	M center in emission	
605-	R center analog of the H <sub>3</sub> molecule	
619-	F' Band " " " negatively charged hydrogen atom in a crystal	Slide 11
650-	Progress in understanding centers	
650-	(Cont) Need for detailed calculations	

### Reel 1 - Track 2

#### The Calculation of the Electronic Structure of Lattice Defects in Ionic Crystals by R. F. Wood (Oak Ridge National Laboratory)

0-7	Introductory Remarks	
8-	Calculations of electronic structure	Slide 1
		Slide 2
50-	Polarization	Slides 3 & 4
	Methods of calculating orthogonization	
110		
185	Localized states	
188	end	

### Reel 1 - Track 3

0	"Creation of Defects in Alkali Halides by Ionizing Radiation" by J. H. Crawford, Jr. Since 1959, it has been accepted that interstitials as well as vacancies were created by ionizing radiation.	
70	Varley mechanism	
95	Kinetic energy	
	Results of studies of mechanisms for radiolysis	
180	Klick and Patterson studies	
200	Conflict between theory and results	
210	Interstitial halogens and the form in which they exist in the crystal. Room temperature experiments. Nadeau and Suzuka experiments with F centers	
240		Slide 8
248		Slide 9
260	Absorbtion F centers - Bleaching time	Slide 10
273	Effect of bleaching on flow stress of irradiated KCl	Slide 11
306		Slide 12
315		Slide 13

- 320 Slide 14  
 363 Effect of adding impurities Slide 15  
 385 We know more about the quantitative aspects than in 1962.  
 The mechanism still remains to be established. Interstitials exist at room  
 temperature - irradiation, but the form in which they exist is unknown.  
 Radiolytic processes at room temperature  
 400 One creation mechanism is needed.  
 Black reaction is responsible for differences

Reel 1 - Track 4

- 7- F- Aggregate Centers in Alkali Halide Crystals by Herbert Rabin  
 (U.S. Naval Research Laboratory)>  
 10 This review is facilitated by a recent survey article written with W.  
 D. Compton.  
 15 Three areas Slide 1  
     1- Experimental Evidence for Models  
     2- Processes of Formation and Destruction  
     3- Theory & Studies Relating to Excited States  
 36- Experimental Evidence for Models Slide 3  
     A- Electric properties - None found  
     B- Magnetic properties - Siedel will discuss this. F2 & F3  
         Models.  
     C- Optical properties -  
 56- Unpolarized bleaching  
 65-  
 M. Veda was the first to use polarized light in examination of the M center.  
 97- M. centers - evaluation of optical studies.  
 120 A Quadratic relation between F and M centers can be expected Slide 5  
 126- KCl Slide 6  
 142- Slide 7  
 150- Slide 8  
 162 KCl Slide 9  
 168- Slide 10  
 176- II Processes of Formation and Destruction Slide 11  
     A. Ionizing Radiation  
         1. Low Temperature Slide 12  
         2. Room Temperature Slide 13  
     B. Optical Conversion  
         1. F>M Conversion  
 232  
 237  
 243 NaFl sensitive to temperature Slide 14  
 Luty

	2. Temporary Products	see Siedel	
273	3. Ionized Centers	Veta	
300			Slide 15
314	C. Thermal Treatment		
319	1. Dissociation		Slide 16
324			Slide 17
338	2. Generation		Slide 18
348	III Theory & Studies Relating to Excited States		Slide 19
	A. M - Center Theory	S. Wang	
381	B. First Excited State		
398			Slide 21
415	C. Higher Excited States	Okamoto, Susman	
432	Optical Studies cover the F - aggregates		
	F2 & F3 models of the M & R centers are established.		
	M1 & M2 centers require further work.		
	Mechanisms of optically exciting the centers need study.		
461	F-aggregate centers in other alkali halides may not perform like KCl		
	Additional bands may appear.		
478	End		

### Reel 21 - Track 1

	"Influence of the Host Lattice on Color Centers" by Franco Bassani (Argonne Laboratory and University of Messina)		
	Electronic Effects on centers		
7-	Three Effects on Centers		
10-	1st		
75-	2nd		
133-	Spin orbit - splitting		
	ground state - excited state transitions		
	(Many calculations are described as they are written on the chalkboard.)		
250	State 4 -A-		2 Slides
262	Three bands in the cesium halides		
	Frequency survey references to the work of others in the field.		
	symmetry		
	splitting of degeneracies.		
349-	Third determination of the position of the energy that is in the color center.		
	calculation of excitation spectra		
	Work of F. Luty		
409	K Band L Band		

435	Excitation states about the continuum
470	Scattering theory shows resonance states in the continuum.
520	Density of states in the continuum.
525	Slide L
570	Kojima
606	Resonance states
644	Shape of absorption band
685	Theoretical Understanding of the L Band
688	End
690-712	Question
713-732	Lins Shape
733-805	Question - L Bands
806-837	polarization energy
838-911	2 questions, excitations - polarization - conduction bands

### Reel 2 - Track 2

"Anionic Impurities in Alkali Halide Crystals" by J. Rolfe (National Research Council of Canada)

	60 papers have been written on this topic since Stuttgart (1962).
28	Classifications of Anionic Impurities Slide 1 soluble impurities
82-110	partially soluble, adventitious impurities
111-117	insoluble or slightly soluble impurities
118	Location of dipolar anionic impurities opening a large new field
150	Expected ionic activity
153	Dr. Rolfe's research on conductivity to measure impurities Slide 2
172	Slide 3
190	Slide 4
206	Slide 5
	Doping with divalent anionic impurities have not produced alkali halides results similar to silver halides.
230	End

### Reel 2 - Track 2

"Rare Earth Impurities in the Alkali Halides" by Walter E. Bron (IBM)

260-	Start
280-	Slide?

295	Slide?	
		crystal field - knowledge of the lattice field
312		Slide 3
340	Effects of lattice motion	
354		Slide 4
	Group Theory	
400	Strong electron lattice couplings	
433	Slide?	
450	KBr spectrum	
	Electron lattice coupling	
470	End	

Reel 2 - Track 3

"Production of Ultrapure Alkali Halides - A Survey" by C. T. Butler  
(Oak Ridge National Laboratory)

1-	We use KCl. Others use KBr	
	We want clean crystals	
	Zone refining	
	Techniques	
	1 - sublimation (Anderson) inferior method	
40-	2 - ion exchange (Fredericks)	
	3 - zoning	
	4 - chemistry	
52	Chemical purification	Slide 1
72	Fusion - filtration tube	
83		Slide 2
134	Clean your furnace	
	keep your laboratory clean	
	clean rooms cost \$200 a sq. ft.	
150	Vacuum tight furnace	
153		Slide 3
180		Slide 4
206		Slide 5
	Need information on physical purity - NaCl	
	Need literature on purity. Lack of literature on best purification or growth techniques.	
245	End.	

Reel 2 - Track 4

"Trapped Hole Centers in the Alkali Halides" by Charles J. Delbecq  
(Argonne National Laboratory)

1-		
17-	Vk (x2) center model	Slide 1
31-	ultraviolet and red	Slide 2
55	spin orbit interaction	
87-	C12 ions in the alkali chlorides	Slide 3
130	trapping of holes - Vk center is the primary configuration.	
135	Hole centers which involve an impurity	
142	H center - Kanzig	
147-193	Review of five models	Slide 1
225-230	Stabilization of Vk centers	
235-245	V centers	
260-300	F & Cl- centers in KCl. interstitials	Slide 4
312	V1 centers	
382	Electron - Hole Recombination Luminescence (Kabler, Wakita, Veta Murray and Keller)	Slide 5
420	Electron - Hole Recombination Luminescence (Yuster, Delbecq, Ghosh, Timusk and Martienssen)	Slide 6
475	tunneling probabilities	Slide 7
495	End.	

Reel 3 - Track 1

"Color Centers in Magnesium Oxide" by John E. Wertz (Minnesota)

7-	Start	impurities	
36	Spectra for F centers in MgO	references to articles	Slide 1
104			Slide 2
114			Slide 3
120			Slide 4
141	Trapped hole centers	Models proposed by Seitz	
170			Slide 5
185			Slide 6
203			Slide 7
205			Slide 8
222			Slide 9
238			Slide 10
250	thermoluminescence		

260		5 slides
278		Slide 16
301	Charge release experiments	Slide 17

Similarities and differences with respect to alkali halides.  
Differences due to divalent positive and divalent negative ion lattice.

318 End.

### Reel 3 - Track 2

"Influence of External Perturbations on Optical Transitions in Electron Excess Centers" by Charles P. Slichter (University of Illinois)

1-	Work on the R center external perturbations have enabled us to spin-orbit splitting	
20	Degeneracy creates two main problems	
	1 - Adiabatic approximation	
	2 - Dilemma of quantization(87)	
40	List of workers in areas covered by this paper	Slide 1
55	Henry & Schnatterly did most of this work	
89		Slide 2
120	spin-orbit splitting	
160		Slide 3
240	quenching orbital angular momentum	Slide 4
284		Slide 5
378		Slide 6
415		Slide 7
421	Energies in the	Slide 8
450	Work by P. R. Moran	
		Slide
	Conclusion from examination of Optical Absorbtion lines	
493	End.	

### Reel 3 - Track 3

"Color Centers in Alkaline Earth Fluorides" by A. Smakula (Crystal Physics Laboratory, M.I.T.)

3-	
50	We have studies CaF <sub>2</sub> , but not Sr F <sub>2</sub> and BaF <sub>2</sub> Potassium, copper and similar



72	Impurities not dangerous in study of color centers.	
77	More dangerous are oxygen in hydrogen	
100	CaF <sub>2</sub> Vienna experiments	
106	Gottingen improved experiments	
112		Slide 2
135	Absorption of crystals additively colored	Slide 3
188	X-rays. Ultraviolet absorption of colored fluoride crystals	Slide 4
217	Influence of impurities on the position of color center bands, absorption coefficients for pure and oxygen contaminated crystals	Slide 5
290	Doping with Yttrium Fluoride analysis of bands	Slide 6
310	Doping with NaF analysis of bands	Slide 7
337	Doping at low temperature	Slide 8
350	" " " "	Slide 9
357		Slide 10
366		Slide 11
	Absorption bands, spectra Far behind alkali halides	
388	End.	

#### Reel 3 - Track 4

"Summary and Concluding Remarks" by H. Pick (Physikalisches Institute, Universitat Gottingen)

17	I attended 1956 conference we have reached peaks in this flat country
35	Survey talks have helped
40	Papers covered: color centers and hole centers in pure, mixed and doped crystals. Aggregate centers and impurity centers with many difference foreign cations, anions, and mutual atoms.
	Theory of optical transitions of zero phonon lines.
45	Techniques and methods High Energy Resonance
54	Three impressions 1 - Concern with number of papers presented 2 - Color center physics is a basic part of solid state physics 3 - Youthful workers

- 74 Important advances in work on the F center  
 85 L Band theory in a preliminary stage  
 96 Zero phonon lines are an excellent means for the control of symmetry parameters  
 and study of the couplings between electronic and vibrational sites.  
 122- F aggregate centers  
 146- Need experimental methods controlling radiation effects  
 150- Vibration effects.  
 Infra-red work  
 165 Impurities and better crystals  
 174 Broader concept of a dynamic theory  
 178 Shift from alkali halides to other materials, particularly those of higher technical  
 applicability.  
 184 Study of ordering forces  
 203 End.

Reel 4

- 0-26 Introductory Remarks by R. J. Maurer  
 27- Historical Background of the Work on Color Centers by R. Hilsch (Physikalisches  
 Institut der Universitat Gottingen)  
 1925 - Student from Stuttgart to Gottingen to work with Prof. R. W. Pohl. Now 81,  
 Pohl retired 10 years ago.  
 48- photoconductivity on diamonds  
 Diamond was destroyed  
 Alkali halides were used  
 measure bands  
 A. Smakula worked on the measurements.  
 Atomic physics and quantum physics were popular  
 Our solid state research was isolated. Having nothing to do with atomic theory, it  
 was scarcely noticed.  
 90- We developed the techniques for the investigation of alkali halides.  
 We used vacuum photoelectric cells with sodium for the ultraviolet.  
 measurement techniques  
  
 Growth crystals. Optical equipment  
 110- Discoloration notices. "F centers" or "color centers" first used in  
 1930.  
 120- Impurities  
 spin resonance  
 At Gottingen work was done to study the alkali halides to apply acquired knowledge  
 to other substances.  
 It has benefitted solid state physics and semi-conductor work.

- 1930 - Baurer. Light reflection reduced by KBr coating.  
Smakula - developed this at Zeiss
- 1932 - F centers in alkali halide crystals migrate at high temperatures. Electrolytic discoloration. Problems remain here.
- 145- 1938 - Controlled migration of F centers  
Solid state amplifier proved practical. No technical device developed.
- 150- 1948 - three electrode Germanium crystals or transistor
- Pohl - preferred facts to theories
- 1946 - F. Seitz review article on Gottingen work.  
They lacked interest in theories  
Experiments were what was needed to furnish interpretations.
- 175- Great number of papers at this meeting  
Problem to organize knowledge of our field.  
Thank program committee
- 205- Abstracts
- 206-210 Maurer concluding remarks.

Previous conferences:

1956 - Argonne

1962 - Stuttgart

1959 - Corvallis

1965 - Urbana